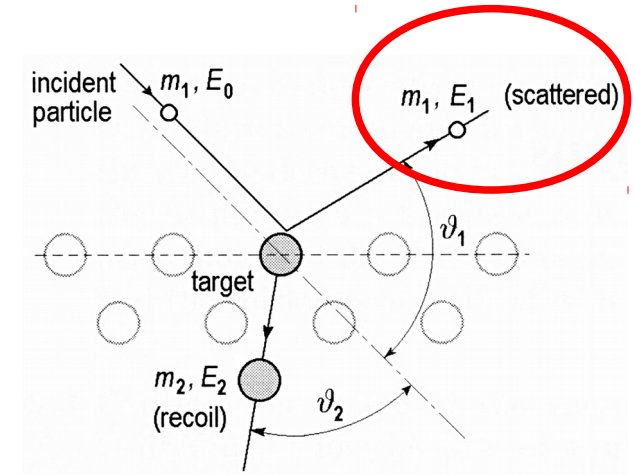
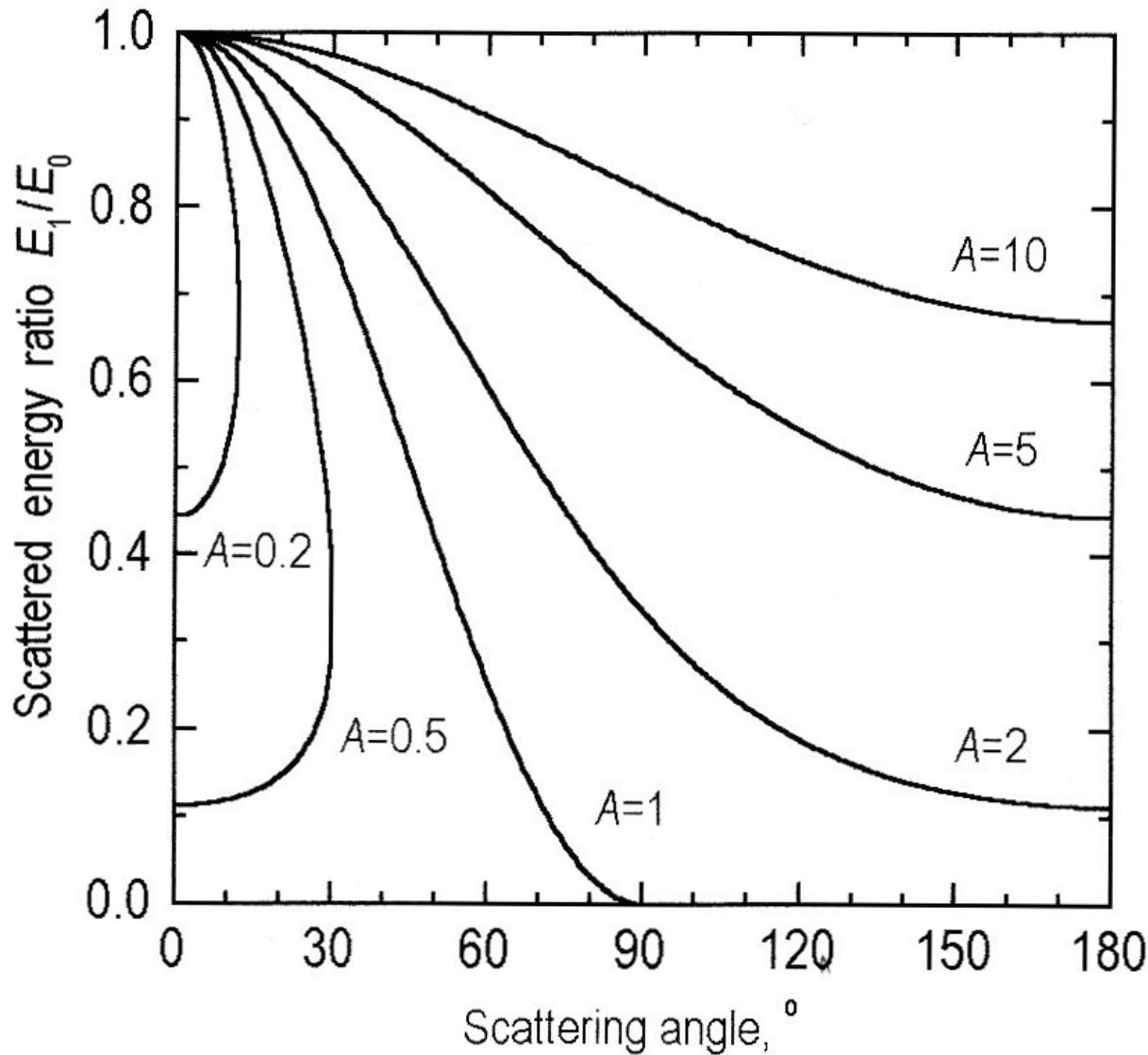


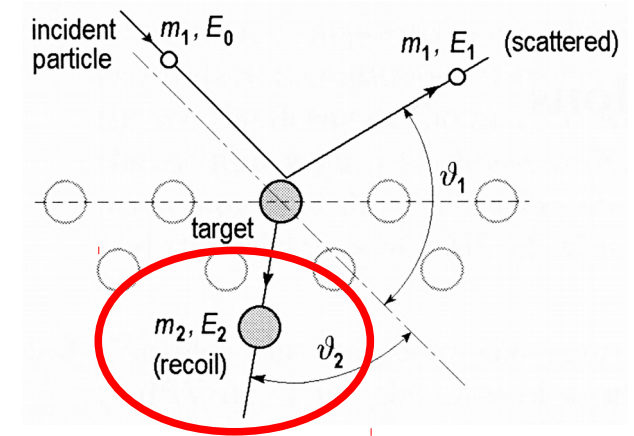
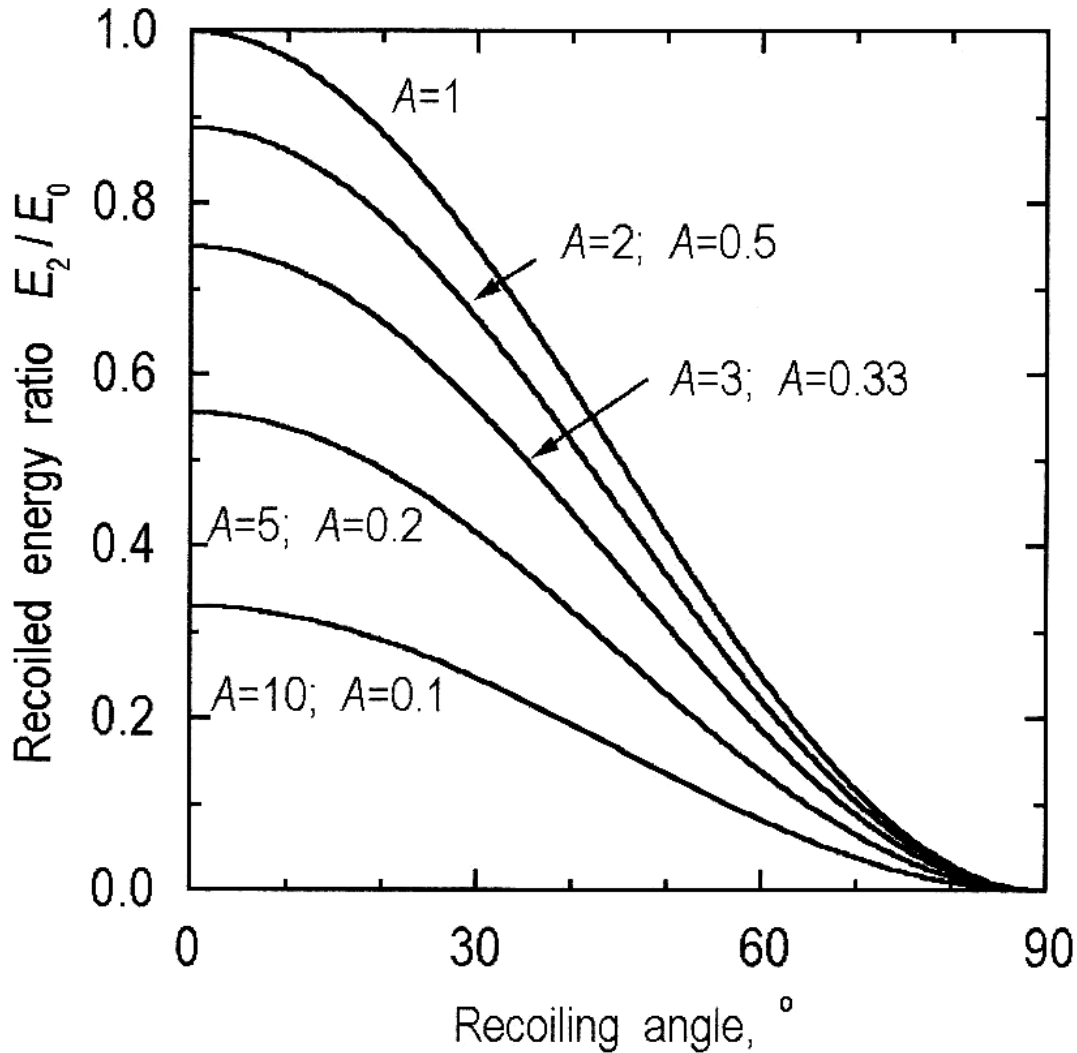
**Fig. 6.1.** Schematic diagram of a binary elastic collision of an incident ion (projectile) with a surface atom (target). The projectile of mass  $m_1$  has initial energy  $E_0$ . The projectile's final scattering angle is  $\vartheta_1$  and its final energy is  $E_1$ . The target particle of the mass  $m_2$ , initially at rest, recoils at an angle  $\vartheta_2$  with energy  $E_2$

# Partikelstoß: Energie des gestreuten Atoms



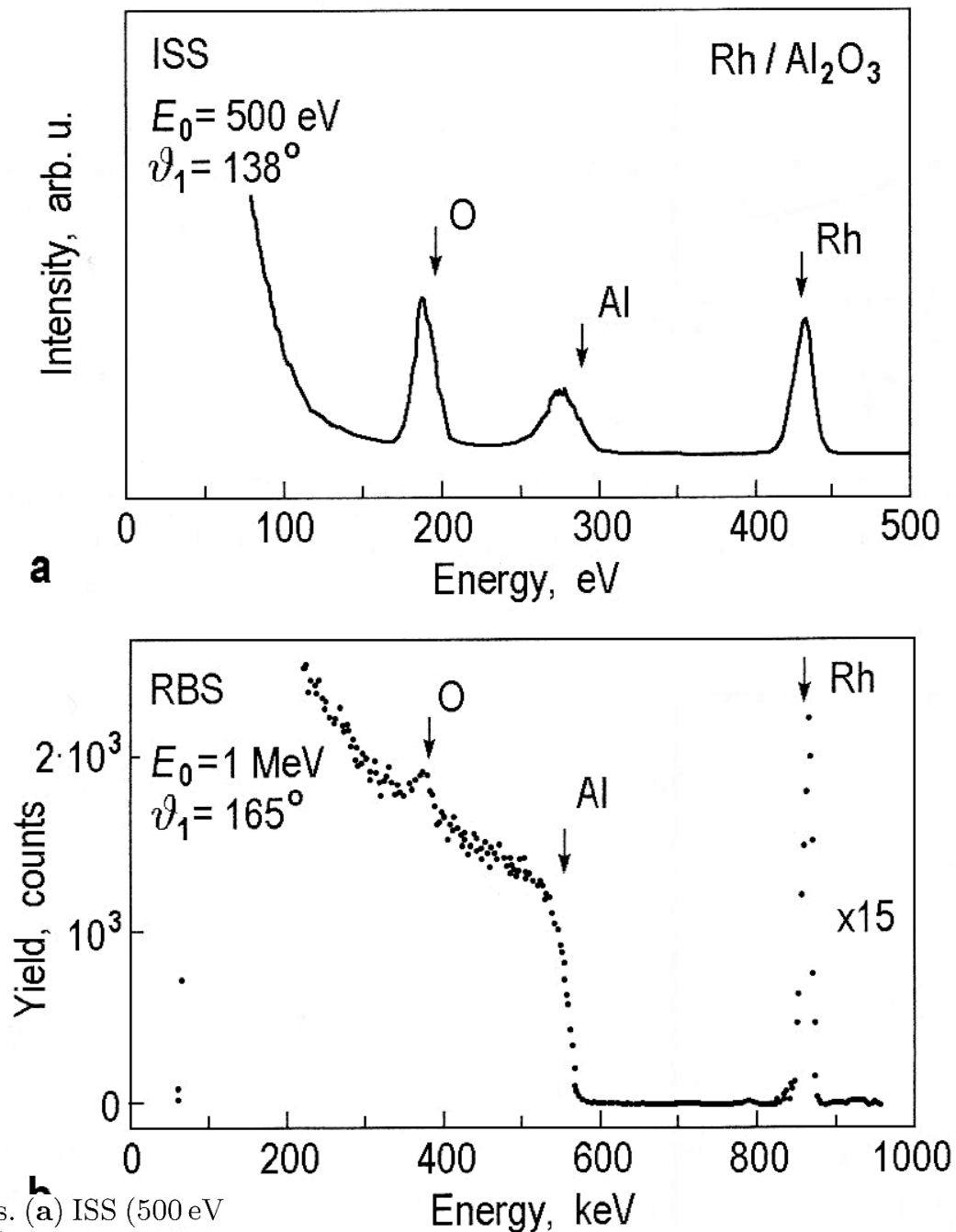
**Fig. 6.2.** Graphical representation of relation (6.4): Energy ratio  $E_1/E_0$  for the scattered particles as a function of the scattering angle  $\vartheta_1$  for various values of the mass ratio  $A = m_2/m_1$

# Partikelstoß: Energie des Rückstoßteilchens



**Fig. 6.3.** Graphical representation of relation (6.5): Energy ratio  $E_2/E_0$  for the recoil particles as a function of the recoiling angle  $\vartheta_2$  for various values of the mass ratio  $A = m_2/m_1$ . The  $E_2/E_0(\vartheta_2)$  curves for  $A$  are identical to those for  $A^{-1}$

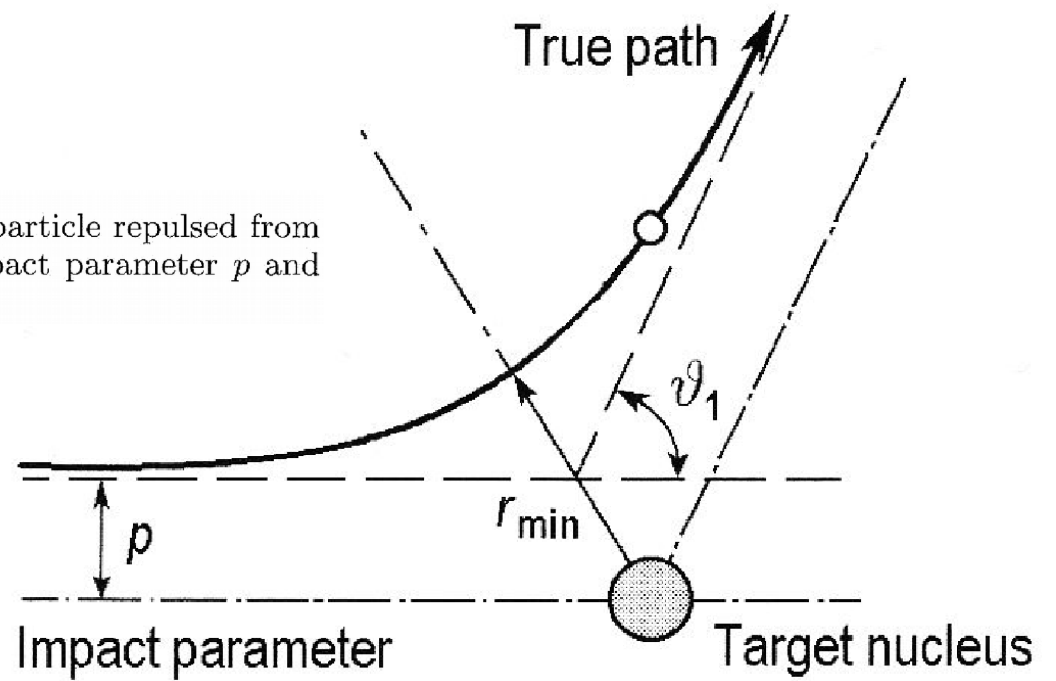
# Elementanalyse



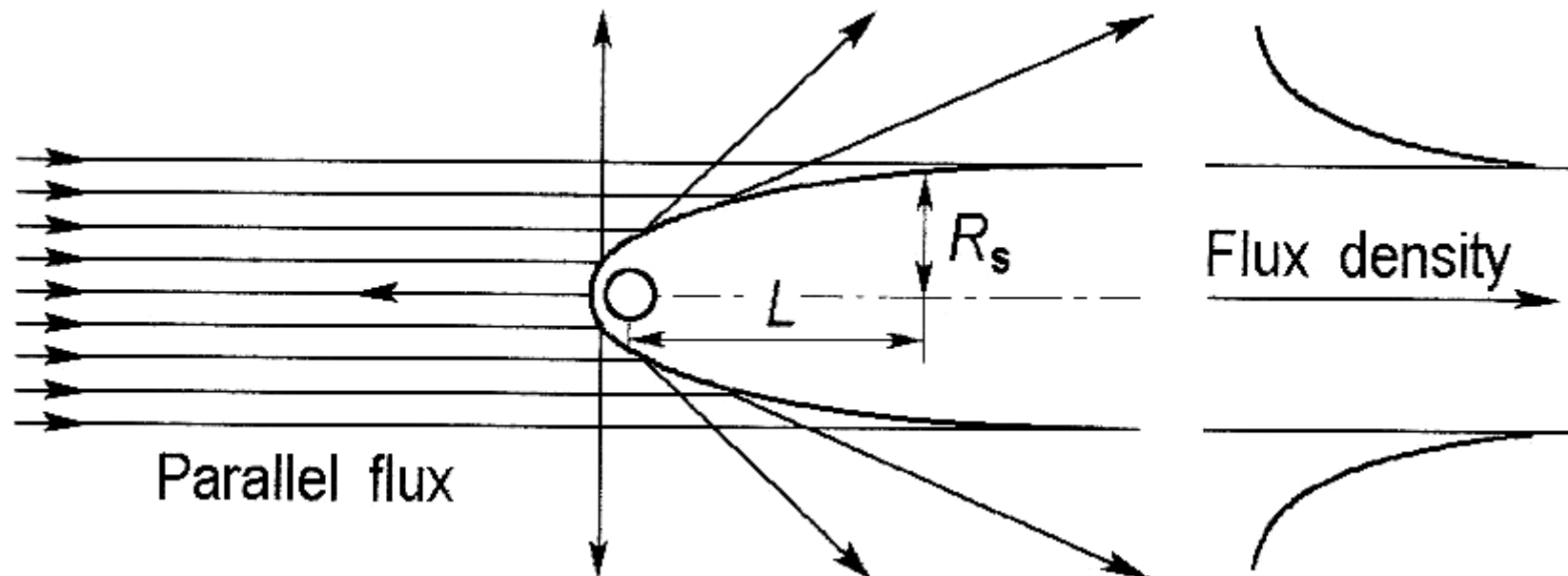
**Fig. 6.4.** Surface elemental analysis using ion scattering techniques. (a) ISS (500 eV He<sup>+</sup>) and (b) RBS (1 MeV He<sup>+</sup>) spectra of a Rh/Al<sub>2</sub>O<sub>3</sub> sample (an alumina film, formed by oxidation in air, with a deposited ~1 ML of rhodium). The arrows indicate the peak positions calculated with the binary collision model (after Linsmeier et al. [6.1])

# Winkelverteilung

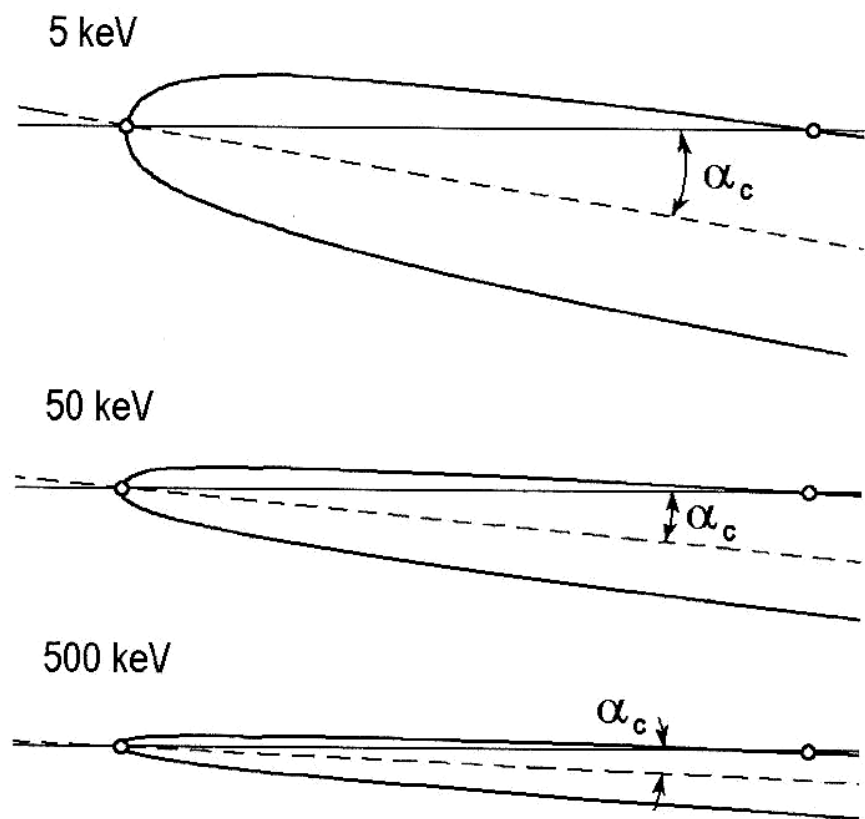
**Fig. 6.5.** A schematic view of the trajectory of a scattered particle repulsed from the target nucleus by the repulsive Coulomb force. The impact parameter  $p$  and the distance of the closest approach  $r_{\min}$  are denoted



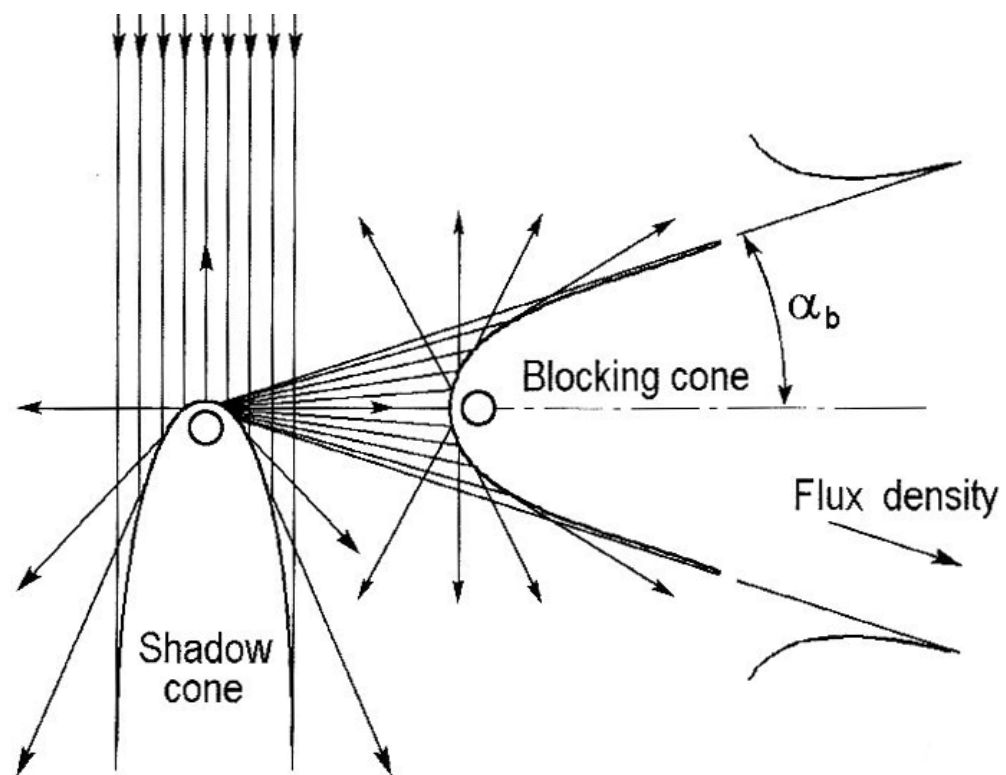
**Fig. 6.6.** Shadow cone formed from trajectories of projectile ions scattered from a target atom



# Shadowing & blocking



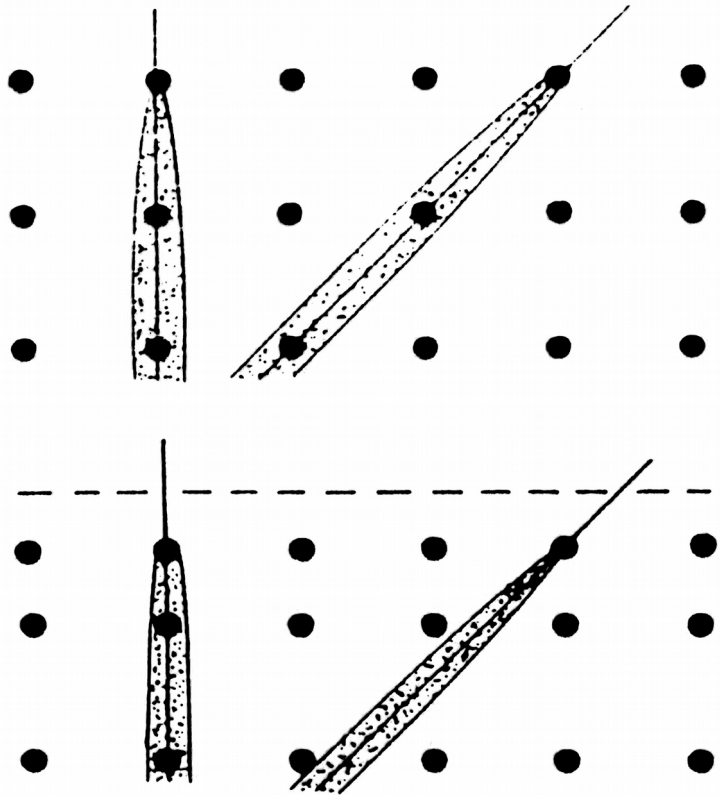
**Fig. 6.7.** Calculated shadow cones for  $\text{Li}^+$  ions with energy of 5 keV, 50 keV, and 500 keV scattering from Ag atoms. The critical angles of shadowing  $\alpha_c$  are indicated. The shadow cone width and critical angle decrease substantially with increasing ion kinetic energy (after Williams [6.3])



**Fig. 6.8.**

# HEIS shadow cones

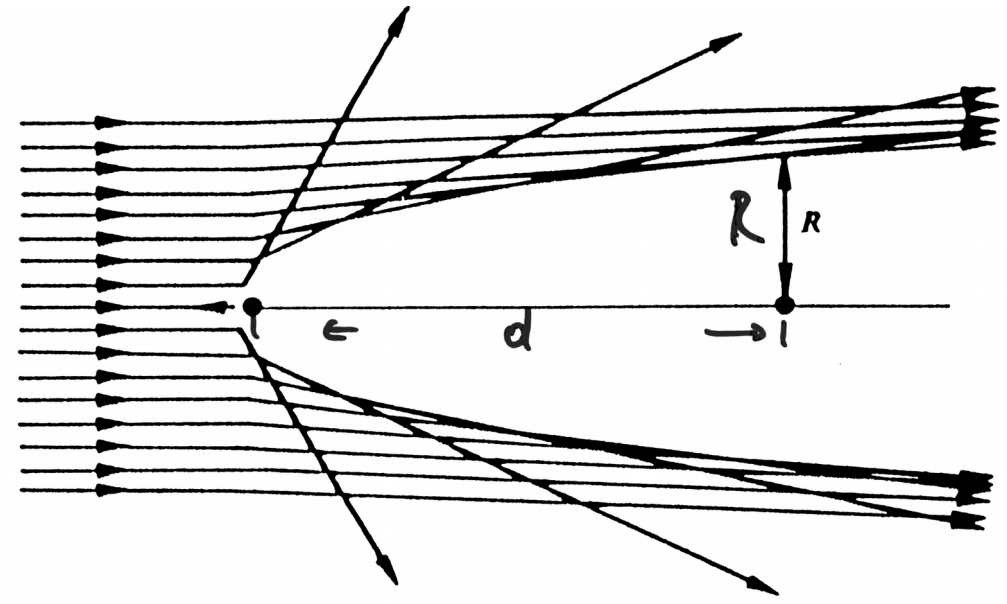
unrelaxed vs. relaxed



inter layer spacings  
modelling → simulation

head-on scattering:

$$E' = E_0 \left( \frac{m_1 - m_2}{m_1 + m_2} \right)^2$$

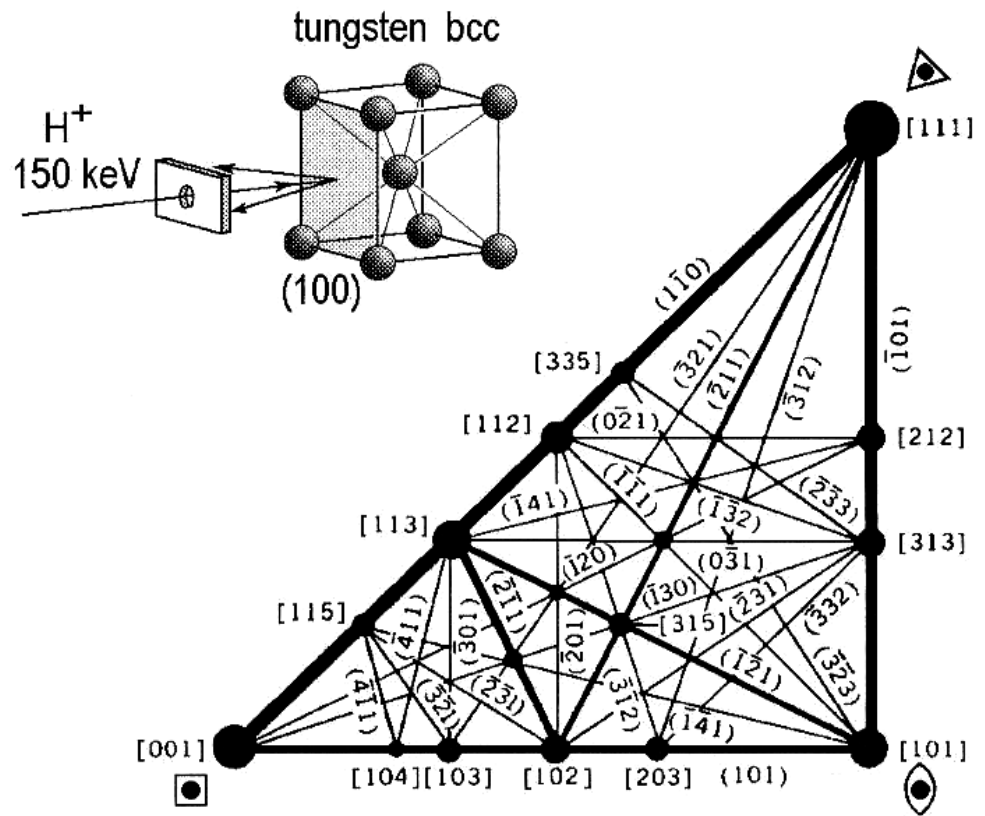
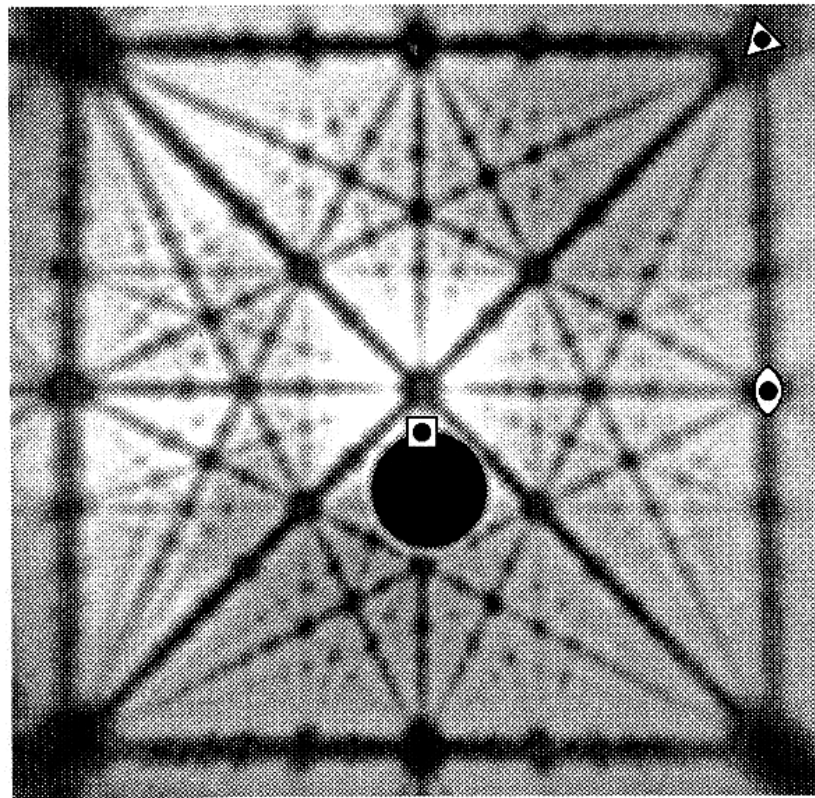


$$R = 2\sqrt{Z_1 Z_2 e^2 d / E_0}$$

$Z_{\text{ion}}$   
 $Z_{\text{TARGET}}$

primary beam  
energy

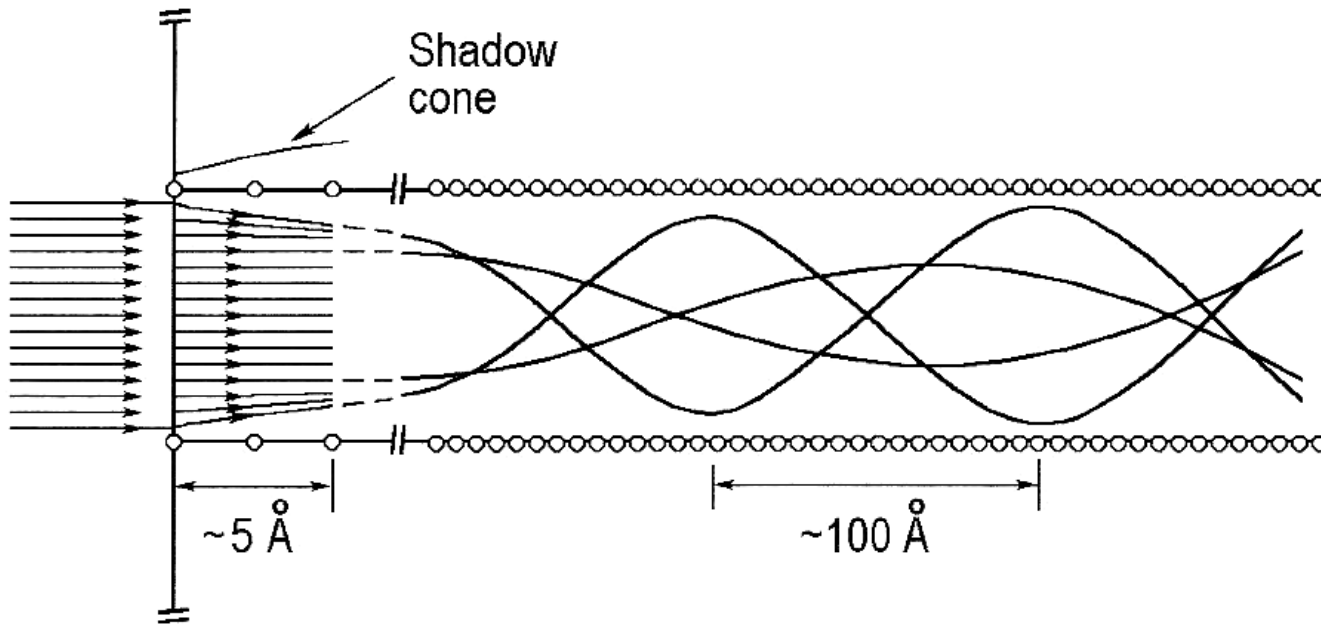
# Blocking pattern of W(100)



**Fig. 6.9.** [100] blocking pattern for 150 keV protons incident on a thick single crystal of tungsten in the non-channeling ( $\sim 5^\circ$ ) direction and backscattered along various high symmetry blocking directions. The angular distribution of scattered protons was monitored with radiation sensitive film. In this print from the film the black lines and spots correspond to proton-deficient regions in the film (After Barrett, Müller and White [6.5])

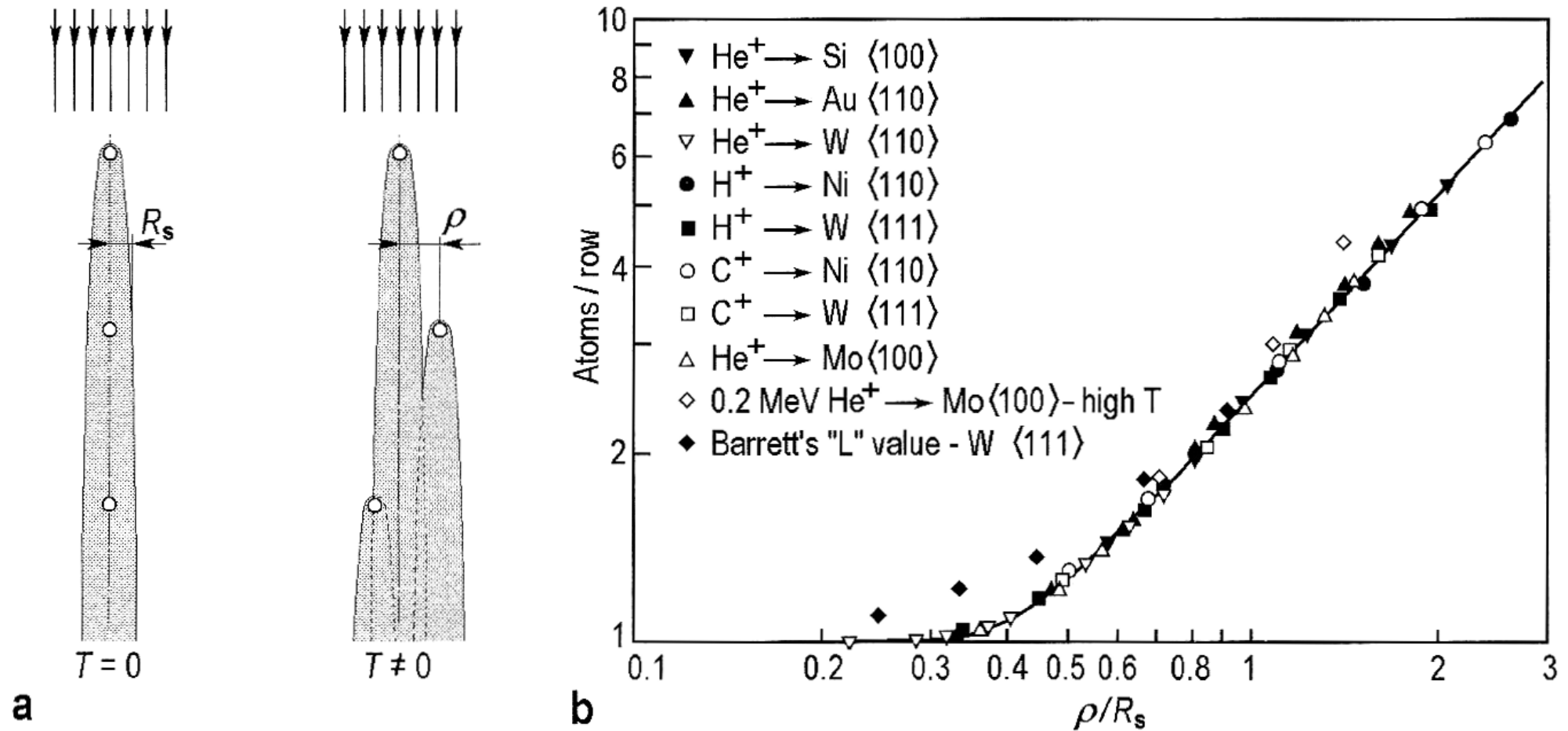


# Channeling



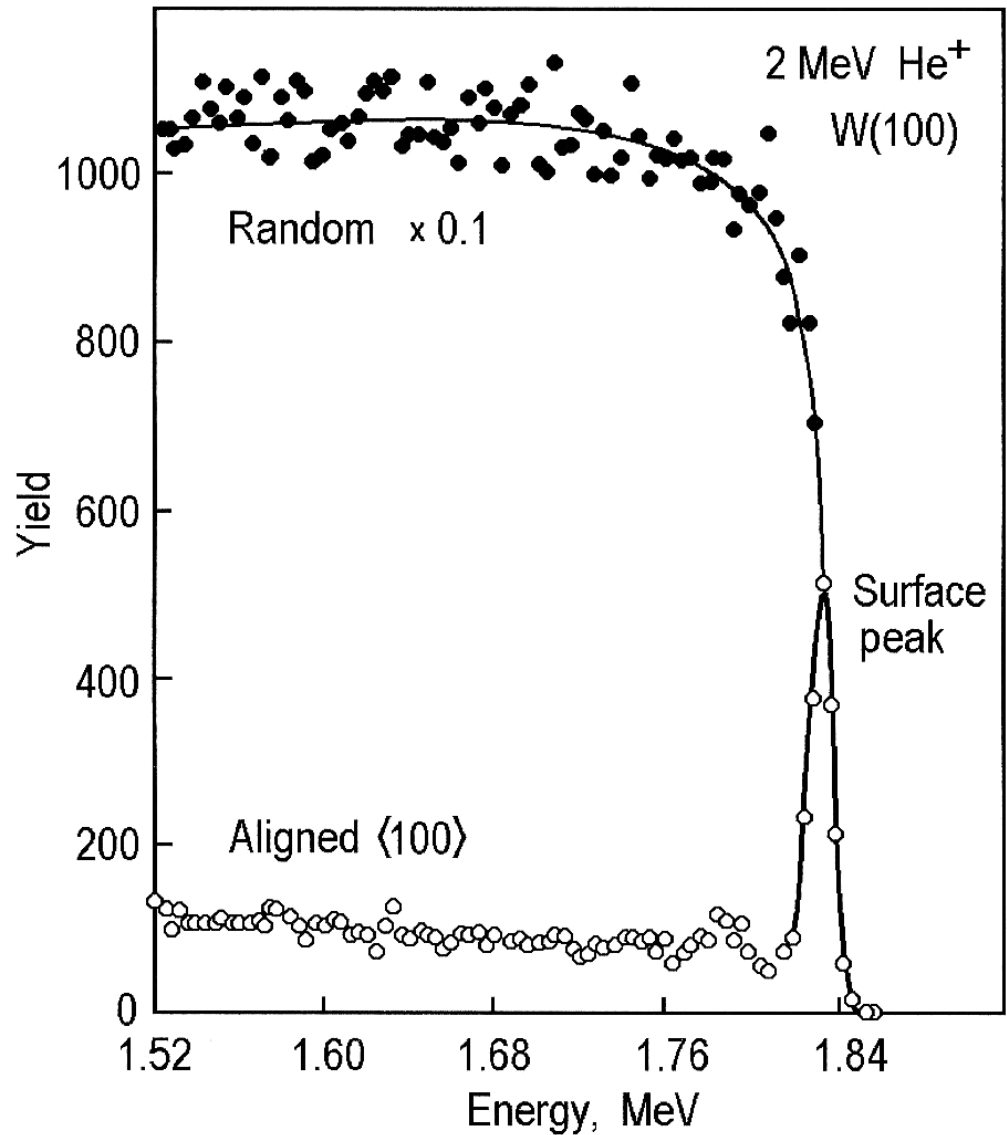
**Fig. 6.10.** Schematic illustration of the particle trajectories undergoing scattering at the surface and channeling within the crystal. The depth scale is compressed relative to the width of the channel in order to display the shape of the trajectories (after Feldman, Mayer and Picraux [6.6])

# Role of Temperature



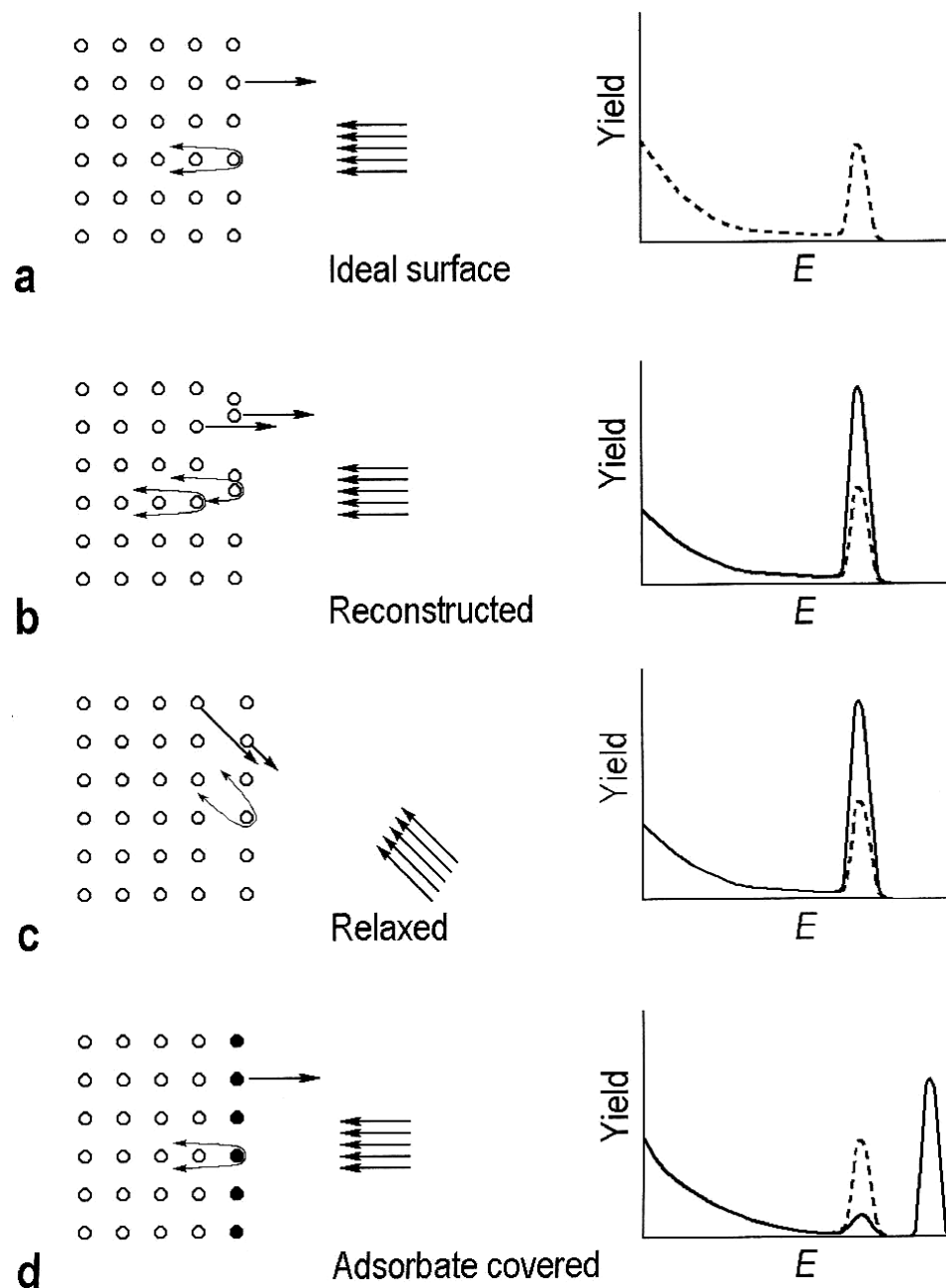
**Fig. 6.23.** (a) Schematic illustration of the shadow effect along an atomic row at zero and non-zero temperature.  $R_s$  indicates the radius of the shadow cone at interatomic distance,  $\rho$  is the amplitude of the thermal atomic vibrations. (b) Number of atoms per row visible to the incident ion beam as a function of  $\rho/R_s$  calculated for a large set of ion-target combinations (after Stensgaard et al. [6.18])

# Surface peak



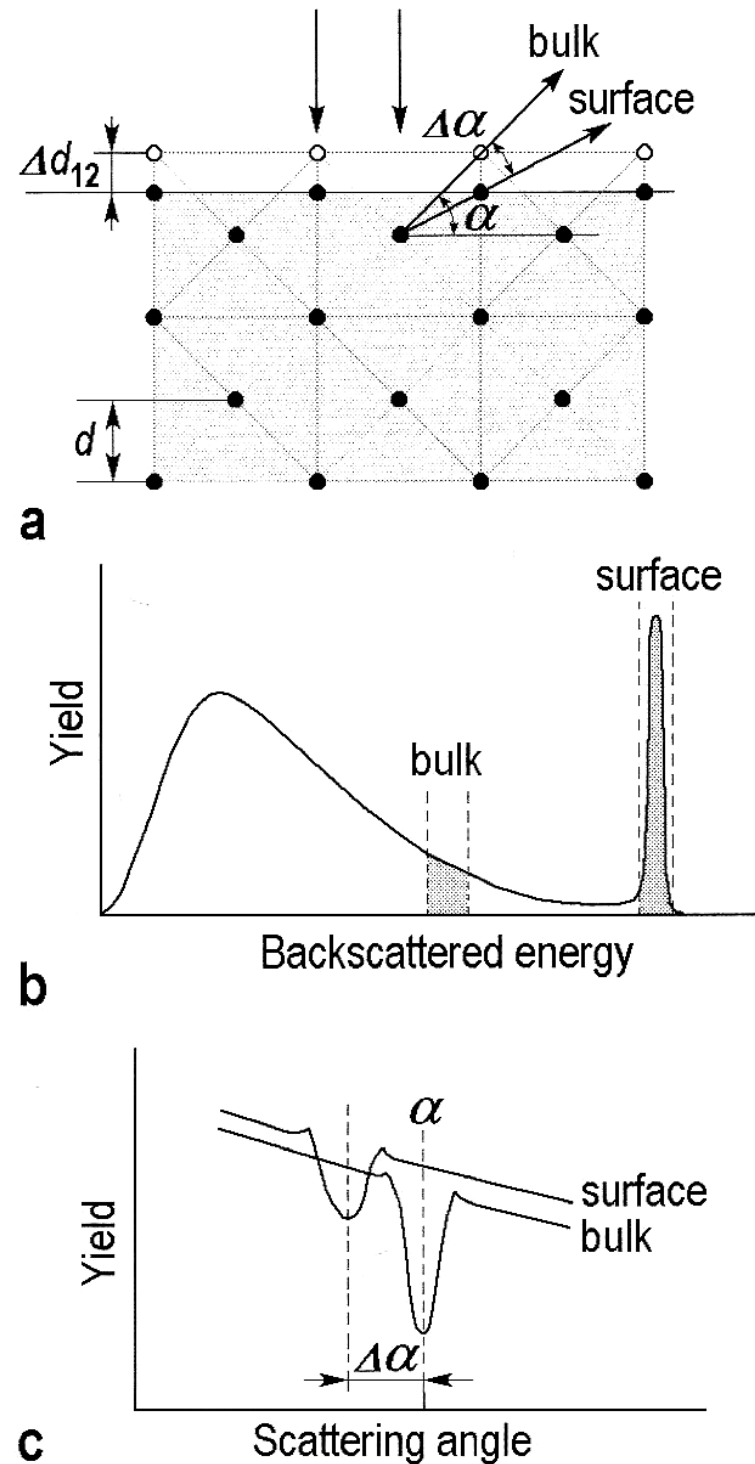
**Fig. 6.24.** RBS spectra for 2 MeV He<sup>+</sup> ions incident on W along the  $\langle 100 \rangle$  axial direction (“aligned” spectrum, open circles) and in the non-channeling direction (“random” spectrum, closed circles) (after Feldman et al. [6.19])

# Idealized backscattering patterns



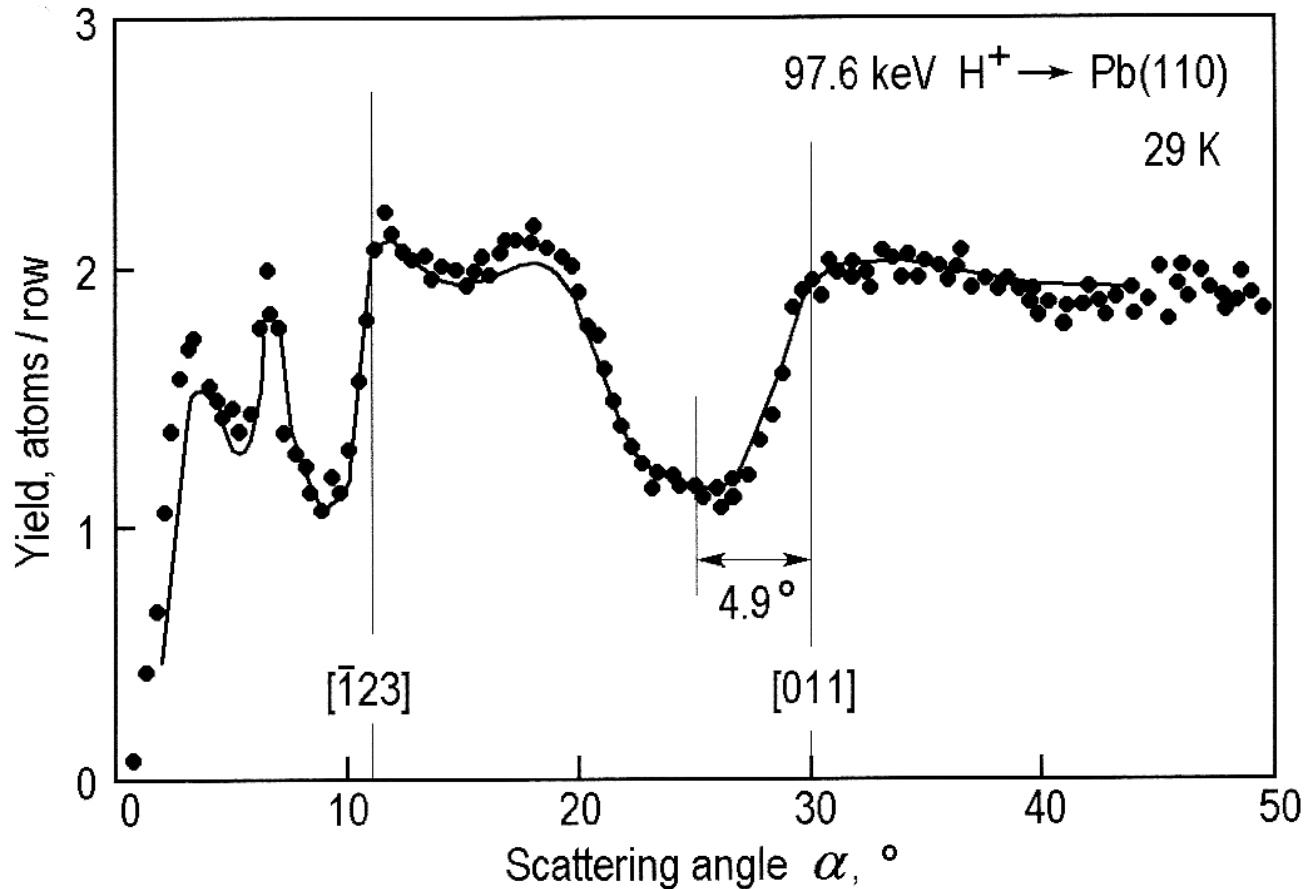
**Fig. 6.25.** Schematic diagram of the expected backscattering signal from various surface structures: (a) ideal crystal surface; (b) surface with lateral reconstruction; (c) surface relaxed in the normal direction; (d) surface with an adsorbate layer (after Feldman et al. [6.6])

# Separating bulk & surface



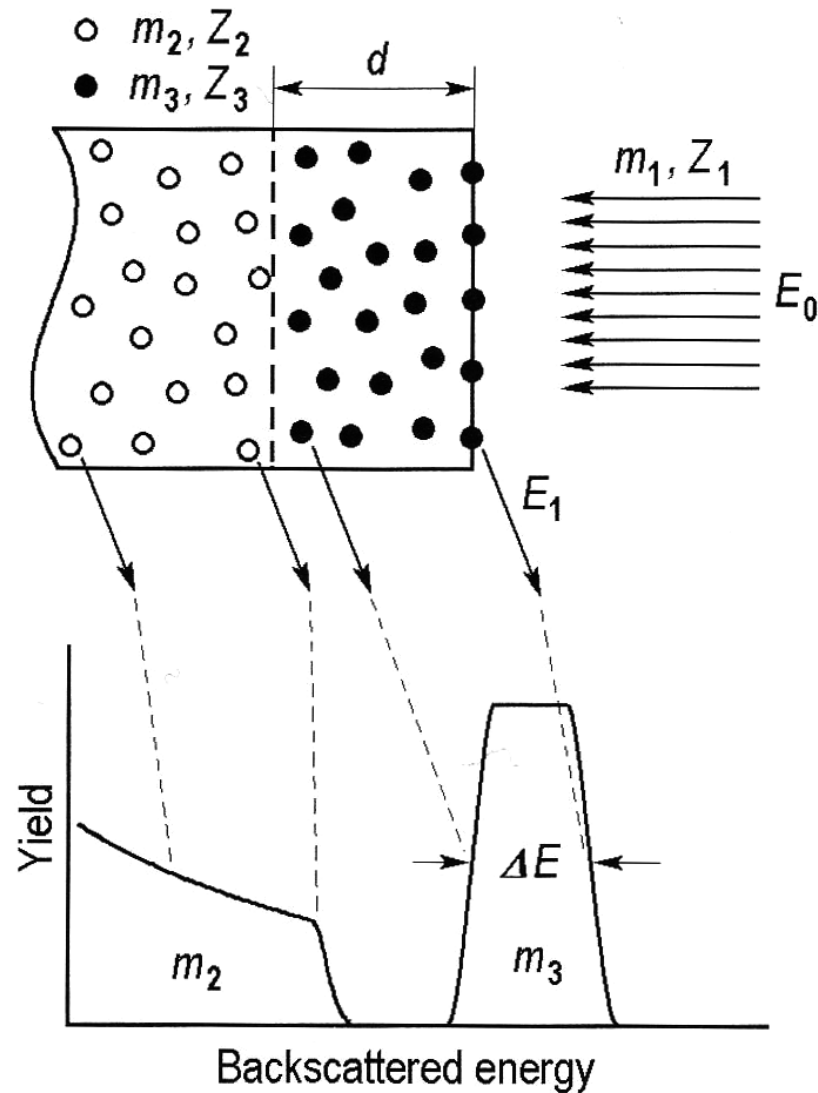
**Fig. 6.26.** Schematic illustration of the method to study surface relaxation. In scheme (a) actual positions of atoms are shown by closed circles, while open circles show the positions of the atoms in the unrelaxed ideal first atomic layer. Due to relaxation by  $\Delta d_{12}$ , the surface blocking minimum direction is tilted by  $\Delta\alpha$  with respect to the bulk axis. In the experiment,  $\Delta\alpha$  is measured by plotting the surface and bulk scattering intensities (b) as a function of scattering angle (c).  $\Delta d_{12}$  is calculated from  $\Delta\alpha$  according to (6.15) (after Turkenberg et al. [6.20])

# A spectacular example: Pb(110)



**Fig. 6.27.** Angular dependence of the surface backscattering peak from a Pb(110)1 $\times$ 1 surface at 29 K. The blocking minimum at 25.1 $^\circ$  is shifted from the bulk [011] direction due to the inward relaxation of the outermost atomic layer of the crystal. The best fit of Monte Carlo simulations (solid curve) is obtained for  $\Delta d_{12}/d = -(17.2 \pm 0.5)\%$  and  $\Delta d_{23}/d = +(8.0 \pm 2.0)\%$  (after Frenken et al. [6.21])

# Layered materials



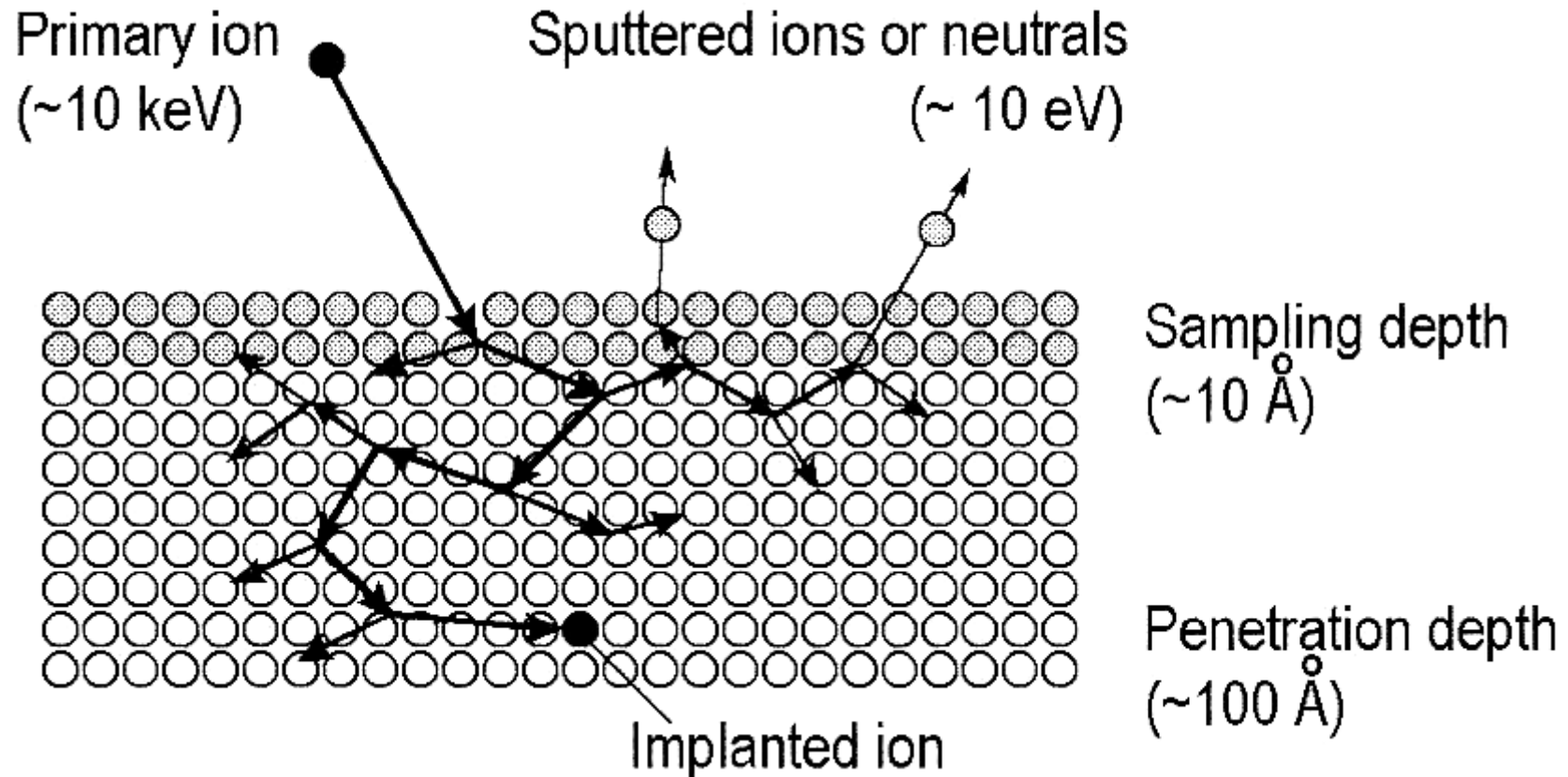
**Fig. 6.28.** Schematic diagram of the energy spectrum of ions ( $m_1, Z_1, E_0$ ) scattered from a sample composed of a substrate ( $m_2, Z_2$ ) and a film ( $m_3, Z_3$ ) of thickness  $d$ . For simplicity, both film and substrate are assumed to be amorphous to neglect the structural effects. (after Feldman et al. [6.6])

Ionen

zerstäuben

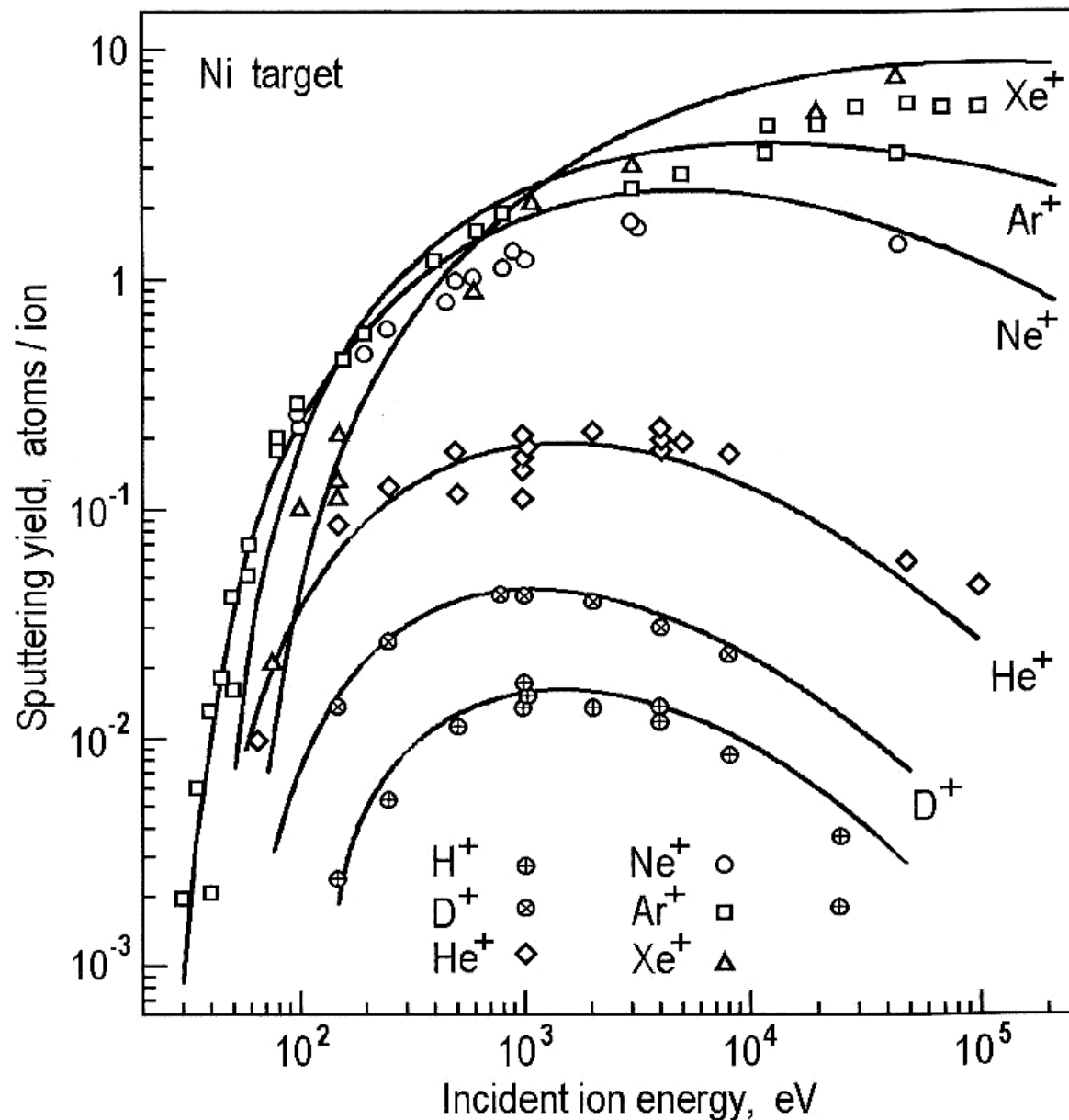


# Ionenzerstäuben (sputtering)



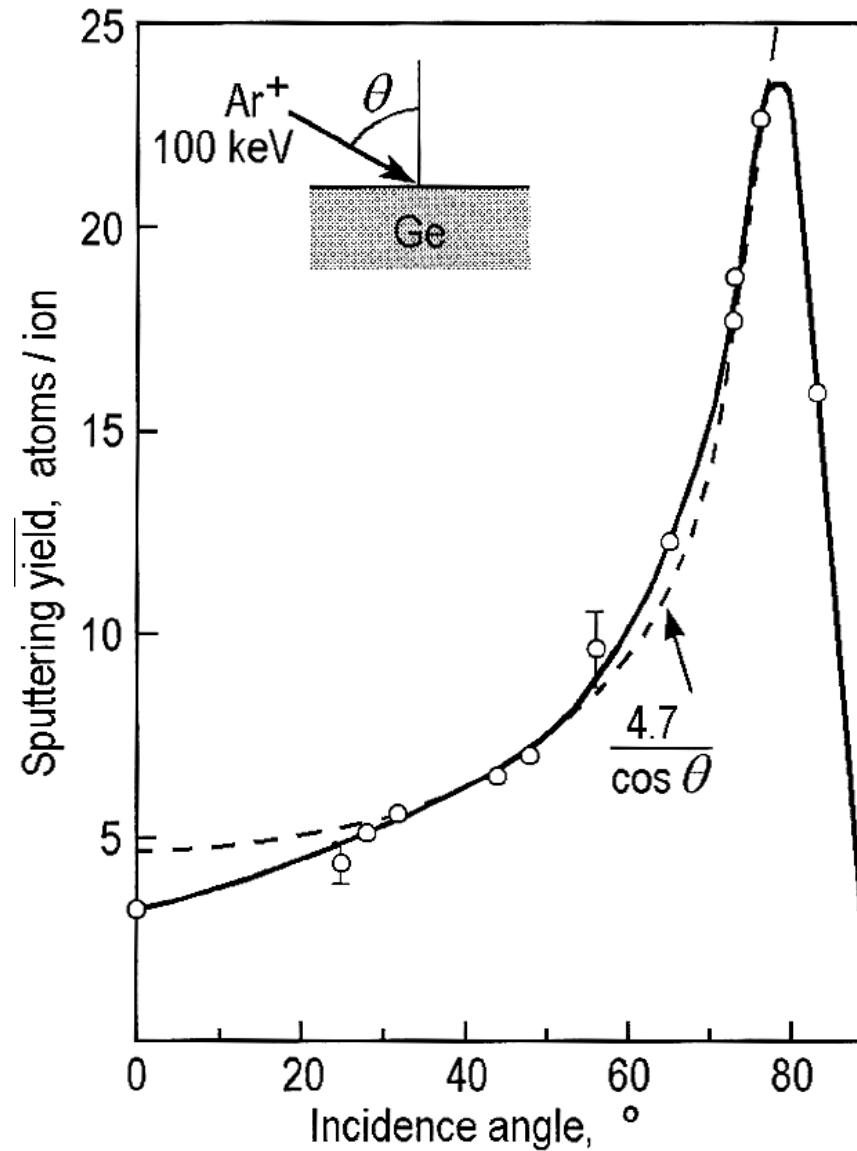
**Fig. 6.11.** Schematic representation of the processes taking place upon penetration of the impinging ion into the solid. The cascade of collisions results in ion implantation and sputtering of surface species. The shown numerical values provide a feeling for the orders of magnitude for the ion penetration depth, escape depth, and energy of sputtered species when bombarding the surface with 10 keV ions

# Ionenzerstäuben: Primärenergie



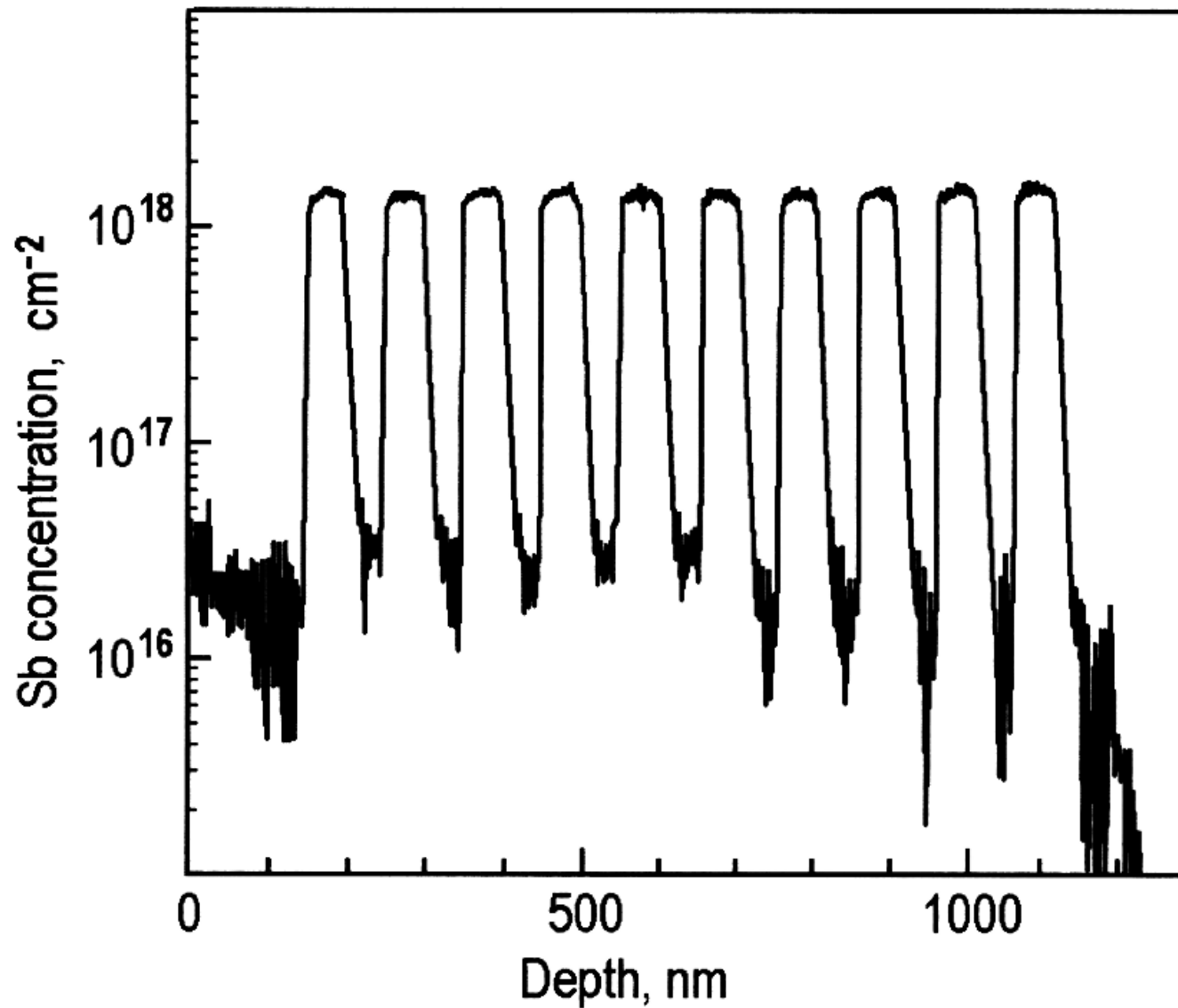
**Fig. 6.12.** Sputtering yield as a function of the primary ion energy for bombardment of a polycrystalline Ni sample by various ions at normal incidence (after Ziegler et al. [6.7])

# Ionenzerstäuben: Einfallswinkel



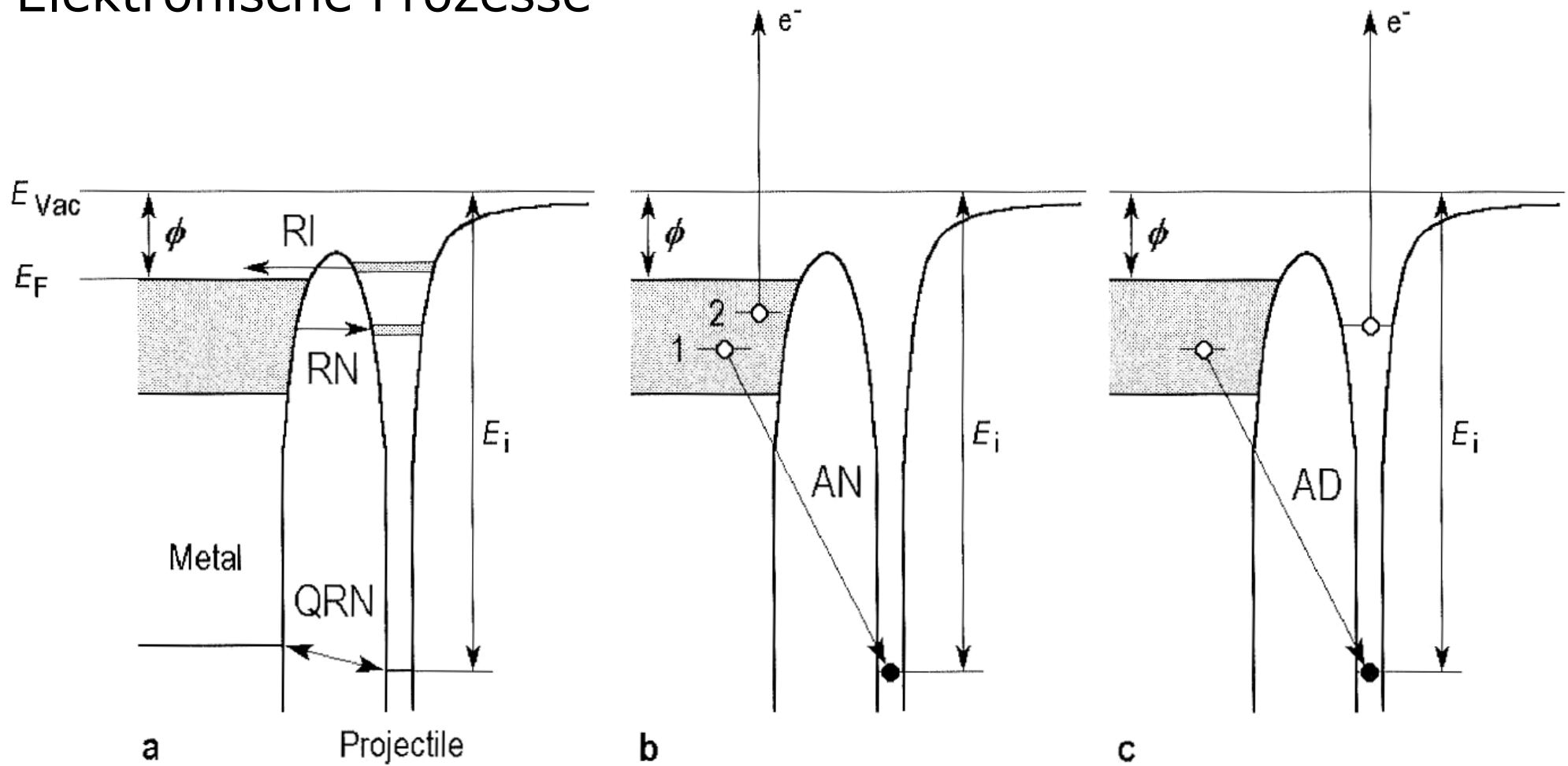
**Fig. 6.13.** Variation of the sputtering yield with ion incidence angle  $\theta$  for 100 keV argon bombardment of germanium which amorphizes readily under ion impact (after Wilson et al. [6.8])

## Secondary Ion Mass Spectrometry (SIMS)



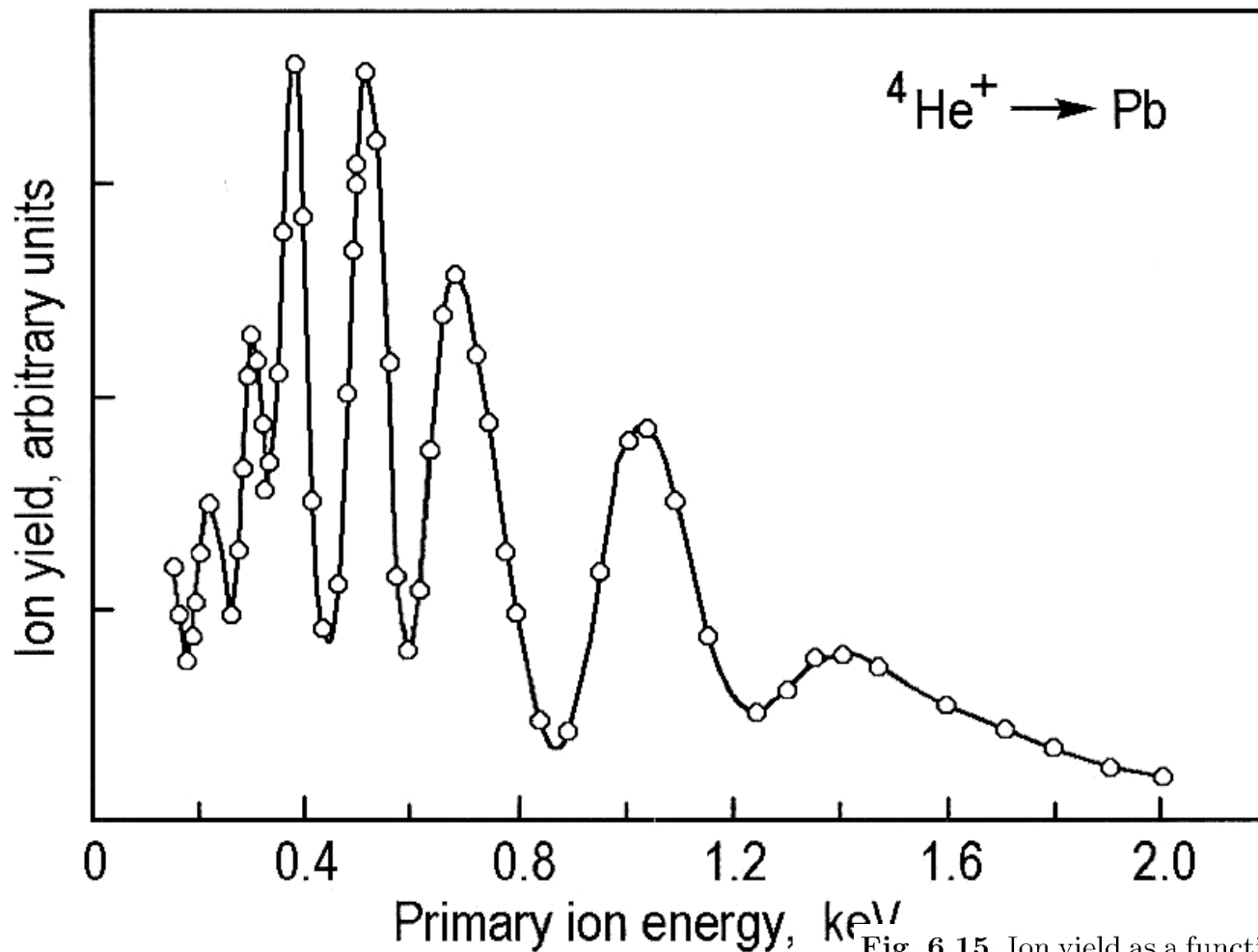
**Fig. 6.31.** SIMS depth profile of Sb in the modulation-doped silicon multilayer structure grown by molecular beam epitaxy (after Casel et al. [6.25])

# Elektronische Prozesse



**Fig. 6.14.** Schematic energy level diagrams showing the charge exchange processes between a solid and an ion. (a) One-electron processes are represented by resonance neutralization (RN), resonance ionization (RI), and quasi-resonant neutralization (QRN). Two-electron processes are represented by (b) Auger neutralization (AN) and (c) Auger de-excitation (AD).  $E_F$  is the Fermi energy of the solid,  $E_{Vac}$  is the vacuum energy,  $\phi = E_{Vac} - E_F$  is the work function of the solid, and  $E_i$  is the ionization energy of the ion (after Hagstrum [6.9])

# Oscillatory neutralization



**Fig. 6.15.** Ion yield as a function of the primary ion energy for  ${}^4\text{He}^+$  scattering from Pb targets. The oscillatory behavior of the ion yield is explained by *quasi-resonant neutralization processes* as follows. Electron exchange processes are dependent upon the interaction time, which is a function of the primary ion velocity (energy). For a sufficiently high ion velocity, the collision time is so short that the electron has just enough time to transfer to the incident ion; the ion then scatters away, neutralized. For a somewhat lower ion velocity, there will be sufficient time for the electron to transfer to the incident ion and back again to its parent atom, resulting in no neutralization. For progressively lower ion velocities, a succession of electron-exchange events takes place. The minima in the ion yield correspond to those ion velocities (collision times) where charge exchange results in scattering of a neutral. The period of oscillation is constant when the data are plotted against inverse velocity of the incident ion (after Erickson and Smith [6.10])